



# Course Specifications

Institution:	<b>Majmaah University.</b>
Academic Department :	<b>Department of chemistry</b>
Programme :	<b>Bachelor degree of chemistry</b>
Course :	<b>Kinetic Chemistry</b>
Course Coordinator :	<b>Ebthag ELhassan</b>
Programme Coordinator :	<b>Dr.Gehan Alaemary</b>
Course Specification Approved Date :	<b>20/12 / 1435 H</b>



## A. Course Identification and General Information

1 - Course title : <b>Kinetic Chemistry</b>	Course Code: <b>Chem 412</b>
2. Credit hours : <b>3(Two Hours Theoretical + Three Hours Workable)</b>	
3 - Program(s) in which the course is offered: <b>Chemistry</b>	
4 – Course Language : <b>Arabic</b>	
5 - Name of faculty member responsible for the course: <b>Ebthag ELhassan</b>	
6 - Level/year at which this course is offered : <b>seven Level</b>	
7 - Pre-requisites for this course (if any) : • <b>Chemistry thermodynamics</b>	
8 - Co-requisites for this course (if any) : <b>Practical course</b>	
9 - Location if not on main campus :( <b>faculty of education Zulfi</b> )	
10 - Mode of Instruction (mark all that apply)	
A - Traditional classroom <input type="checkbox"/>	<input checked="" type="checkbox"/> What percentage? <input type="checkbox"/> <b>80 %</b> <input type="checkbox"/>
B - Blended (traditional and online) <input type="checkbox"/>	<input type="checkbox"/> What percentage? <input type="checkbox"/> <b>0 %</b> <input type="checkbox"/>
D - e-learning <input type="checkbox"/>	<input type="checkbox"/> What percentage? <input type="checkbox"/> <b>0 %</b> <input type="checkbox"/>
E – Correspondence <input type="checkbox"/>	<input type="checkbox"/> What percentage? <input type="checkbox"/> <b>0 %</b> <input type="checkbox"/>
F - Other <input type="checkbox"/>	<input checked="" type="checkbox"/> What percentage? <input type="checkbox"/> <b>20 %</b> <input type="checkbox"/>
Comments : ..... <input type="checkbox"/>	

## B Objectives

What is the main purpose for this course? <b>Study the Rate and Classification Of Chemical Reaction</b> <b>Determination Of Rate Of Chemical Reaction</b> <input type="checkbox"/>
Briefly describe any plans for developing and improving the course that are being implemented : <b>The use of interactive whiteboard teaching instead of the chalkboard.</b> <b>use of the Web in modern additions to the course .</b> <input type="checkbox"/>





## C. Course Description

### 1. Topics to be Covered

List of Topics	No. of Weeks	Contact Hours
Definitions for kinetic Chemistry	1	4
The rate of reaction	1	2
Kinetics of particles	1	2
The order of a chemical reaction	1	2
Law of speed of reaction	2	4
Measuring the order of reaction	1	2
The applications of types of order of reaction	3	8
Complex interactions	2	4
Effect of temperature	1	2
Activation energy	1	2
Theories that explain the occurrence of chemical reactions	2	4
<u>Practical</u>		
Measure the speed of chemical reaction (first order, second order)	3	6
effect of concentration on the speed of reaction , determined the order of reaction	2	4
Effect of temperature on the speed of reaction, Measuring activation energy	2	4

### 2. Course components (total contact hours and credits per semester):

<input type="checkbox"/>	Lecture	Tutorial	Laboratory	Practical	Other:	Total
Contact Hours	30	---	30	---	---	60
Credit	30	---	15	---	---	45



3. Additional private study/learning hours expected for students per week.

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4. Course Learning Outcomes in NQF Domains of Learning and Alignment with Assessment Methods and Teaching Strategy

	NQF Learning Domains And Course Learning Outcomes	Course Teaching Strategies	Course Assessment Methods
<b>1.0</b>	<b>Knowledge</b>		
1.1	<b>Know the Rate and Classification Of Chemical Reaction</b>	lecture discussion, mutual dialogue	Oral tests at the beginning of each lecture, Written tests, final examination
1.2	<b>Determination Of Rate Of Chemical Reaction</b>		
1.3	<b>Study Rate Of Chemical Reaction and The Factor which affected On It</b>		
1.4	<b>Study Of Mechanism of Reversible and Irreversible Kinetic Reaction</b>		
1.5	<b>Theories that explain the occurrence of chemical reactions</b>		
1.6			
<b>2.0</b>	<b>Cognitive Skills</b>		
2.1	<b>Application of The Chemical Operation To Link between The Theoretical and Workable Material</b>	problems, Laboratory study Open discussions	Continuous questions- duties - practical test
<b>3.0</b>	<b>Interpersonal Skills &amp; Responsibility</b>		
3.1	<b>Dealing with team spirit in experiments</b>	Working in groups within the lab Collective seminars	Oral questions, Correct experimental results
3.2	<b>Creating constructive competitive spirit</b>		
3.3	<b>Encourage communication between students</b>		





	<b>NQF Learning Domains And Course Learning Outcomes</b>	<b>Course Teaching Strategies</b>	<b>Course Assessment Methods</b>
<b>4.0</b>	<b>Communication, Information Technology, Numerical</b>		
<b>4.1</b>	<b>Development of communication skills</b>	<b>Problems research, study discussion</b>	<b>Oral and written exercises Follow-up practical books,</b>
<b>4.2</b>	<b>Development of numerical skills</b>		
<b>4.3</b>	<b>Use chemical Internet sites and doing some calculation</b>		
<b>5.0</b>	<b>Psychomotor</b>		
<b>5.1</b>	<b>Mastering laboratory experiments</b>	<b>Practical course</b>	<b>Follow-up practical books,</b>

### 5. Schedule of Assessment Tasks for Students During the Semester:

	<b>Assessment task</b>	<b>Week Due</b>	<b>Proportion of Total Assessment</b>
<b>1</b>	<b>Questions and exercises</b>	<b>fourth and fifth</b>	<b>10%</b>
<b>2</b>	<b>Theoretical midterm exam</b>	<b>sixth</b>	<b>20%</b>
<b>3</b>	<b>practical midterm exam</b>	<b>eighth</b>	<b>20%</b>
<b>4</b>	<b>Final practical exam.</b>	<b>fourteenth</b>	<b>20%</b>
<b>5</b>	<b>Final Theoretical exam</b>	<b>Last week</b>	<b>40%</b>





## **D. Student Academic Counseling and Support**

**Two hours of weekly academic guidance**

## **E. Learning Resources**

### **1. List Required Textbooks :**

**Chemistry electrical electrolytic conductivity Ahmed Abdulaziz Al Owais**

### **2. List Essential References Materials :**

- **Foundations of physical chemistry, Adel Ahmed Jrare**
- .....

### **3. List Recommended Textbooks and Reference Material :**

- **Chemistry electrical electrolytic conductivity Ahmed Abdulaziz Al Owais**

### **4. List Electronic Materials :**

- **.Wikipedia**
- .....

### **5. Other learning material :**

- **Power point - CD show**



## **F. Facilities Required**

### **1. Accommodation**

- **Prepared Classroom with Interactive whiteboard**
- **40 chair.**

### **2. Computing resources**

- **Laptop special for Professor only**
- .....

### **3. Other resources**

- **There is a need to equip lab special for this course**

## **G Course Evaluation and Improvement Processes**

**1 Strategies for Obtaining Student Feedback on Effectiveness of Teaching:**





<ul style="list-style-type: none"> <li>• Meeting with the students academic excellence and the stumble</li> <li>• Identification of evaluation for the course form students</li> </ul>
<b>2 Other Strategies for Evaluation of Teaching by the Program/Department Instructor :</b> <ul style="list-style-type: none"> <li>• Benefit from the expertise of the members of the section</li> <li>• Identify assessment for teachers</li> <li>• Report of the expert from College matchups</li> </ul>
<b>3 Processes for Improvement of Teaching :</b> <ul style="list-style-type: none"> <li>• Courses for Faculty members</li> <li>• Workshop to improve methods of evaluation .....</li> <li>• .....</li> </ul>
<b>4. Processes for Verifying Standards of Student Achievement</b> <ul style="list-style-type: none"> <li>• The patch is checked by faculty member</li> </ul>
<b>5 Describe the planning arrangements for periodically reviewing course effectiveness and planning for improvement :</b> <ul style="list-style-type: none"> <li>• discussion the members section regularly to improve the course</li> <li>• feedback processes for course quality</li> </ul>

**Course Specification Approved**  
**Department Official Meeting No ( ..... ) Date ... / ... / ..... H**

<p><b>Course's Coordinator</b> <input type="checkbox"/></p> <p><b>Name :</b> <input type="checkbox"/> Ebthag Elhassan <input type="checkbox"/></p> <p><b>Signature :</b> <input type="checkbox"/> .....</p> <p><b>Date :</b> <input type="checkbox"/> ... / ... / ..... H <input type="checkbox"/></p> <p><input type="checkbox"/></p>	<p><b>Department Head</b> <input type="checkbox"/></p> <p><b>Name :</b> <input type="checkbox"/> .....</p> <p><b>Signature :</b> <input type="checkbox"/> .....</p> <p><b>Date :</b> <input type="checkbox"/> ... / ... / ..... H</p>
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