



Institution: Majmaah University

Academic Department : Chemistry Programme : Chemistry

Course: Instrumental Analysis Chemistry

Course Coordinator: Lecturer. Amneh Shtaiwi Programme Coordinator: Dr. Gehan Al-Omayri

Course Specification Approved Date: 18/12/1435 H



A. Course Identification and General Information

1 - Course title : Instrumental Analy Chemistry.	ysis Course Code	: Chem 411.			
2. Credit hours: 4					
3 - Program(s) in which the course	e is offered: Chemistr	ry			
4 – Course Language : Arabic					
5 - Name of faculty member respon	nsible for the course:	Lecturer. Amneh Shtaiwi			
6 - Level/year at which this course	is offered: 9 I e	vel			
7 - Pre-requisites for this course (ifQuantitative analytical chemistry					
8 - Co-requisites for this course (if any): • Instrumental Analysis Chemistry lab.					
9 - Location if not on main campus:					
main campus					
10 - Mode of Instruction (mark all	that apply)				
A - Traditional classroom	What percentage?	%			
B - Blended (traditional and online) What percentage? 75 %					
D - e-learning √ What percentage? 25%					
E - Correspondence What percentage? %					
F - Other What percentage? %					
Comments:					

B Objectives

What is the main purpose for this course?

Identification and analysis of electrical equipment, including for analysis methods and Potential Calomtry, Voltamitry and Ampirometery. As well as ways to give entrance to the spectroscopy, chromatography and identify their kinds and principles.

Briefly describe any plans for developing and improving the course that are being implemented:

Using Internet in the research work.





C. Course Description

1. Topics to be Covered

List of Topics	No. of Weeks	Contact Hours
-A general introduction in the electrolytic methods include Potentiometric, colomtric and gravimetric analysis and Electrolytic.	4	12
- Ampirometric and voltamitric titration.	4	12
- Introduction to the spectral analysis methods include visible spectroscopy.	2	6
- Methods of molecular spectroscopy.	1	3
- Methods of atomic spectroscopy.	1	3
Introduction to the Chromatography and distribution coefficient.	1	3
Chromatographic methods of separation sheets, columns and gas chromatography.	2	6

2. Course components (total contact hours and credits per semester):

	Lecture	Tutorial	Laboratory	Practical	Other:	Total
Contact Hours						2
Credit	3		2			4

3. Additional private study/learning hours expected for	
students per week.	••••••





4. Course Learning Outcomes in NQF Domains of Learning and Alignment with Assessment Methods and Teaching Strategy

	NQF Learning Domains And Course Learning Outcomes	Course Teaching Strategies	Course Assessment Methods
1.0	Knowledge		
1.1	 To learn the electrolytic methods include Potentiometric, colomtric and gravimetric analysis and Electrolytic. To learn methods used in the expression of different concentrations, Equilibrium and the formation of complexes. 	Lecture Exercises Discussion	Exams, Questions
1.2	Describe the spectral analysis methods include visible spectroscopy.	Experiments Discussion	Experimental Study, Exams
2.0	Cognitive Skills		
2.1	- Describe the Methods of molecular spectroscopy. - Describe the Methods of atomic spectroscopy	Lecture, Exercises Discussion Lecture,	Exams, Questions Exams,
	Describe the National of months speed oscopy	Exercises Discussion	Questions
3.0	Interpersonal Skills & Responsibility		
3.1	Teamwork	Divide in the form of practical sets.	Oral exercises.
4.0	Communication, Information Technology, Numeri	cal	
4.1	- Calibrations calculations for neutralization interactions , redox , sedimentation and complexes	Lecture, Discussion	Oral exercises Exams.
5.0	Psychomotor		
5.1	Experimental work	Lecture, Discussion	Oral exercises Exams.

5. Schedule of Assessment Tasks for Students During the Semester:

Accessment tack	Week Due	Proportion of Total
Assessment task	Week Due	Assessment





1	First Exam	6	15
2	Second Exam	10	15
3	Final Exam	14	60
4	Resaearch	9	10



D. Student Academic Counseling and Support

Academic Advising

E. Learning Resources

1. List Required Textbooks:

- Instrumental Analysis Chemistry, Ibrahim Al-Zamel. 1993.
- Quantitative analytical chemistry, 5th edition by j.S. Fritz and G.H. Schneck. 1987.

2. List Essential References Materials:

- Instrumental Analysis Chemistry, Ibrahim Al-Zamel. 1993.
- Key creativity in Chemistry, Omar Helwah.

3. List Recommended Textbooks and Reference Material:

• Quantitative analytical chemistry, 5th edition by j.S. Fritz and G.H. Schneck. 1987

4. List Electronic Materials:

chemix, chemsketch, chemdraw programs.

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5. Other learning material :

Crocodile program.

F. Facilities Required

1. Accommodation

Seats and computers.

2. Computing resources

Lap top.





3. Other resources

• Projector.

G Course Evaluation and Improvement Processes

1 Strategies for Obtaining Student Feedback on Effectiveness of Teaching:

- Questionnaires Evaluation.
- 2 Other Strategies for Evaluation of Teaching by the Program/Department Instructor:
 - Discussions.
- **3 Processes for Improvement of Teaching:**
 - Review course plans periodically and adjuste..
- 4. Processes for Verifying Standards of Student Achievement
 - Corrected tests with the teaching staff of the department..
- 5 Describe the planning arrangements for periodically reviewing course effectiveness and planning for improvement :
 - Questionnaires Evaluation.

Course Specification Approved	
Department Official Meeting No () Date /	/ H

Cours	e's Coordinator	Department Head		
Name :	Amneh Shtaiwi	Name :		
Signature :	Amneh Shtaiwi	Signature :		
Date :	18/ 12 / 1435 <i>H</i>	Date :	// H	

